

# A convenient synthesis of CF<sub>3</sub>-containing alkynyl imines by a copper-free Sonogashira coupling of imidoyl chlorides with terminal alkynes

Muwang Chen<sup>a</sup>, Qiaochu Zhang<sup>b</sup>, Xingguo Zhang<sup>a\*</sup>, Ping Zhong<sup>a</sup> and Maolin Hu<sup>a</sup>

<sup>a</sup>College of Chemistry and Materials Engineering, Wenzhou University, Wenzhou 325027, P.R. China

<sup>b</sup>College of Chemistry and Biological Engineering, Yichun University, Yichun 336000, P.R. China

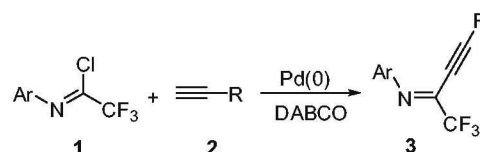
A series of *N*-aryltrifluoroacetimidoyl chlorides and 1-alkynes were converted into  $\alpha$ -imino alkynes by a Sonogashira reaction using Pd(PPh<sub>3</sub>)<sub>4</sub>/DABCO as the catalyst. The reaction proceeded smoothly to give the coupled products in good to excellent yields.

**Keywords:** *N*-aryltrifluoroacetimidoyl chlorides,  $\alpha$ -imino alkynes, Sonogashira reaction

The introduction of a trifluoromethyl group into organic compounds is attractive due to their unique chemical,<sup>1</sup> physical, and biological properties.<sup>2</sup> The trifluoromethylated organic, inorganic, and organometallic compounds have been used in the fields of pharmaceutical, agrochemical, and polymer chemistry.<sup>3,4</sup> Accordingly, the development of new reagents and improved techniques for the synthesis of the trifluoromethyl-containing organic compounds continues to be an important area of research in organofluorine chemistry.<sup>5</sup> Extensive studies have been made to develop a cheaper and more efficient synthetic methodology for the introduction of the trifluoromethyl group into organic compounds.<sup>6</sup>

Trifluoromethyl alkynyl imines can serve as important building blocks for the synthesis of different heterocycles, such as quinolines,<sup>7</sup> pyrazoles and pyrimidines.<sup>8</sup> There are many approaches to the preparation of alkynyl imines.<sup>9–11</sup> Among them, the most straightforward and powerful is the Sonogashira coupling reaction of imidoyl halides with terminal alkynes using Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> and CuI as the catalyst.<sup>12–14</sup> However, the presence of CuI can result in the formation of some Cu(I) acetylides *in situ* which can readily produce an oxidative homocoupling reaction of the alkynes.<sup>15</sup> This side reaction is problematic when the terminal acetylenes are difficult to obtain. To overcome this drawback, examples have been reported concerning palladium-catalysed Sonogashira coupling reaction using copper-free reaction conditions.<sup>16,17</sup>

It is well known that the *sp*<sup>2</sup> carbon-iodine bond is more reactive than carbon-chlorine bond in coupling reaction. Uneyama *et al.*<sup>18</sup> reported the preparation of trifluoromethyl



**Scheme 1**

alkynyl imines from the Sonogashira reaction of *N*-aryltrifluoroacetimidoyl iodides with 1-alkynes catalysed by PdCl<sub>2</sub>/CuI.<sup>18</sup> They also pointed that the corresponding chlorides were recovered intact in the same conditions. Herein, we would like to report our research result of copper-free Sonogashira reaction of *N*-aryltrifluoroacetimidoyl chlorides with 1-alkynes in the catalyst system of Pd(PPh<sub>3</sub>)<sub>4</sub>/DABCO.

The cross-coupling of *N*-(*p*-tolyl)-2,2,2-trifluoroacetimidoyl chloride (**1a**) with phenyl acetylene was chosen as the model reaction to optimise reaction conditions. DABCO was used as the base because triethylamine would evaporate at 110 °C. In screening the catalyst, Pd(OAc)<sub>2</sub>, Pd(OOCCF<sub>3</sub>)<sub>2</sub>, Pd<sub>2</sub>(dba)<sub>3</sub> and Pd(PPh<sub>3</sub>)<sub>4</sub> were tested at room temperature or 110 °C in toluene. As shown in Table 1, Pd(PPh<sub>3</sub>)<sub>4</sub> displayed higher catalytic activity than others. More importantly, increasing the reaction temperature accelerated the reaction to give 93% yield (entry 8). Note that 10 mol % catalyst was necessary and the yield of **3aa** decreased to 70% in the presence of only 5 mol % Pd(PPh<sub>3</sub>)<sub>4</sub> (entry 9). In addition, the use of 1, 4-dioxane, acetonitrile or THF to replace toluene as solvent

**Table 1** Optimisation of Sonogashira reaction of *N*-(*p*-tolyl)-2, 2, 2-trifluoroacetimidoyl chloride with phenylacetylene

Entry	Catalyst/mol% <sup>a</sup>	Solvent	Temperature/°C	<b>3aa</b> Yields/% <sup>b</sup>
1	Pd(OAc) <sub>2</sub> (10)	Toluene	25	41
2	Pd(OOCCF <sub>3</sub> ) <sub>2</sub> (10)	Toluene	25	36
3	Pd <sub>2</sub> (dba) <sub>3</sub> (10)	Toluene	25	30
4	Pd(PPh <sub>3</sub> ) <sub>4</sub> (10)	Toluene	25	48
5	Pd(OAc) <sub>2</sub> (10)	Toluene	110	65
6	Pd(OOCCF <sub>3</sub> ) <sub>2</sub> (10)	Toluene	110	54
7	Pd <sub>2</sub> (dba) <sub>3</sub> (10)	Toluene	110	59
8	Pd(PPh <sub>3</sub> ) <sub>4</sub> (10)	Toluene	110	93
9	Pd(PPh <sub>3</sub> ) <sub>4</sub> (5)	Toluene	110	70
10	Pd(PPh <sub>3</sub> ) <sub>4</sub> (10)	1,4-dioxane	100	43
11	Pd(PPh <sub>3</sub> ) <sub>4</sub> (10)	MeCN	80	41
12	Pd(PPh <sub>3</sub> ) <sub>4</sub> (10)	THF	65	53

<sup>a</sup>Reaction conditions were: **1a** (1.0 mmol), **2a** (1.2 mmol), DABCO(3.0 mmol) in solvent (3 ml).

<sup>b</sup>Isolated yield.

\* Correspondent. E-mail: [xzg@wzu.edu.cn](mailto:xzg@wzu.edu.cn)

afforded lower yield (entries 10–12). Therefore, the catalytic system of Pd(PPh<sub>3</sub>)<sub>4</sub>/DABCO in toluene was considered to be an efficient catalytic system for the Sonogashira cross-coupling of *N*-aryltrifluoroacetimidoyl chlorides. The results are summarised in Table 1.

With the optimised reaction conditions in hand, a series of *N*-aryltrifluoroacetimidoyl chlorides were converted to corresponding trifluoromethyl-containing alkynyl imines. The results are summarised in Table 2. All *N*-aryltrifluoroacetimidoyl chlorides reacted smoothly in the procedure and the products were isolated in excellent yields. Substituent effects of *N*-aryl have significant influence on yields. Electron-donating group in *N*-aryl (e.g. methyl or methoxy groups) raised the yields slightly (entries 1–7). On the contrary, electron-withdraw group in *N*-aryl reduced yields (entries 8–12). The strong electron-withdrawing group on the *N*-aryl (e.g. NO<sub>2</sub>) reduced the stability of the coupling products. This resulted in the decomposition of the products and reduction of the yields<sup>19</sup> (entry 15). In addition, *o*-substituents on the aryl ring reduced the yields (entries 13–15), due to the steric hindrance in the substrate.

In conclusion, we have successfully developed a method for the synthesis of trifluoromethylated  $\alpha$ -imino alkynes by the cross-coupling reaction of *N*-aryl-trifluoroacetimidoyl chlorides with 1-alkynes catalysed by Pd(PPh<sub>3</sub>)<sub>4</sub>/DABCO system.

## Experimental

<sup>1</sup>H NMR were determined in a CDCl<sub>3</sub> solution with a Bruker Avance300 (300 M Hz) spectrometer using tetramethylsilane as the internal standard. <sup>19</sup>F NMR spectra were also obtained in a CDCl<sub>3</sub> solution on Bruker Avance300 (282.2 MHz) spectrometer using CFC1<sub>3</sub> as external standard. All chemical shifts ( $\delta$ ) were expressed in parts per million, and coupling constants (*J*) were given in Hertz. Column chromatography was performed using EM Silica gel 60 (300–400 mesh)

*General procedure for the palladium-catalysed coupling reaction of phenylacetylene 2a with N-(p-Tolyl)-2, 2, 2-trifluoroacetimidoyl chloride 1a*

*N*-(*p*-Tolyl)-2, 2, 2-trifluoroacetimidoyl chloride **1a** (221 mg, 1 mmol) and phenylacetylene **2a** (122 mg, 1.2 mmol) were added successively. To a mixture of Pd(PPh<sub>3</sub>)<sub>4</sub> (115 mg, 0.1 mmol), 1, 4-diazabicyclo[2.2.2]octane (DABCO) (336 mg, 3 mmol) in toluene (5 ml), the reaction mixture was stirred at 110 °C for 8 hours under N<sub>2</sub> atmosphere, then the solvent was evaporated under reduced pressure. The residue

was purified by flash column chromatography on a silica gel using EtOAc/petroleum ether(1:10) as eluant to give the product.

The physical and spectra data of all compounds are as follows.

*4-Methyl-N-(1,1,1-trifluoro-4-phenylbut-3-yn-2-ylidene)aniline (3aa)*: Yellowish oil. <sup>1</sup>H NMR:  $\delta$  7.42–7.47 (m, 3H), 7.32–7.39 (m, 4H), 7.24–7.27 (m, 2H), 2.41 (s, 3H); <sup>19</sup>F NMR:  $\delta$  71.36(s); IR (neat, cm<sup>-1</sup>): 3041, 2928, 2199, 1605, 1499, 1445, 1342, 1151; MS (ESI) *m/z* 288 (M + H<sup>+</sup>); Anal. Calcd for C<sub>17</sub>H<sub>12</sub>F<sub>3</sub>N: C, 71.07; H, 4.21; N, 4.88. Found: C, 71.03; H, 4.19; N, 4.89%.

*Methyl-N-(1,1,1-trifluoro-5-phenylpent-3-yn-2-ylidene)aniline (3ab)*: Yellowish solid, m.p. 95–96 °C <sup>1</sup>H NMR:  $\delta$  8.15 (d, *J* = 8.3 Hz, 1H), 8.11 (s, 1H), 7.64 (d, *J* = 8.3 Hz, 1H), 7.53 (s, 1H), 7.13–7.42 (m, 5H), 4.48 (s, 2H), 2.56 (s, 3H); <sup>19</sup>F NMR:  $\delta$  71.64(s); IR (KBr, cm<sup>-1</sup>): 3043, 2928, 2211, 1592, 1505, 1460, 1373, 1151; MS (ESI) *m/z* 302 (M + H<sup>+</sup>); Anal. Calcd for C<sub>18</sub>H<sub>14</sub>F<sub>3</sub>N: C, 71.75; H, 4.68; N, 4.65. Found: C, 71.72; H, 4.65; N, 4.68%.

*4-Methyl-N-(1,1,1-trifluoro-3-yn-2-ylidene)aniline (3ac)*: Yellowish solid, m.p. 90–92 °C <sup>1</sup>H NMR:  $\delta$  8.11 (d, *J* = 8.6 Hz, 1H), 7.83 (s, 1H), 7.63 (d, *J* = 8.6 Hz, 1H), 7.53 (s, 1H), 3.11 (t, *J* = 7.9 Hz, 2H), 2.60 (s, 3H), 1.72–1.82 (m, 2H), 1.45–1.53 (m, 2H), 1.01 (t, *J* = 7.4 Hz, 3H); <sup>19</sup>F NMR:  $\delta$  67.39(s); IR (KBr, cm<sup>-1</sup>): 3043, 2928, 2864, 2211, 1593, 1505, 1460, 1373, 1154; MS (ESI) *m/z* 268 (M + H<sup>+</sup>); Anal. Calcd for C<sub>15</sub>H<sub>16</sub>F<sub>3</sub>N: C, 67.40; H, 6.03; N, 5.24. Found: C, 67.36; H, 6.05; N, 5.22%.

*4-Methyl-N-(1,1,1-trifluoro-3-yn-2-ylidene)aniline (3ad)*: Yellowish solid, m.p. 84–86 °C <sup>1</sup>H NMR:  $\delta$  8.11 (d, *J* = 8.6 Hz, 1H), 7.83 (s, 1H), 7.61 (d, *J* = 8.6 Hz, 1H), 7.53 (s, 1H), 3.11 (t, *J* = 7.9 Hz, 2H), 2.60 (s, 3H), 1.76–1.81 (m, 2H), 1.50–1.54 (m, 2H), 1.38–1.44 (m, 2H), 0.97 (t, *J* = 7.4 Hz, 3H); <sup>19</sup>F NMR:  $\delta$  67.38(s); IR (KBr, cm<sup>-1</sup>): 3043, 2928, 2864, 2211, 1593, 1505, 1460, 1373, 1154; MS (ESI) *m/z* 282 (M + H<sup>+</sup>); Anal. Calcd for C<sub>16</sub>H<sub>18</sub>F<sub>3</sub>N: C, 68.31; H, 6.45; N, 4.98. Found: C, 68.28; H, 6.42; N, 4.97%.

*4-Methoxy-N-(1,1,1-trifluoro-4-phenylbut-3-yn-2-ylidene)aniline (3ba)*: Yellowish oil. <sup>1</sup>H NMR: 7.48–7.56 (m, 4H), 7.43–7.49 (m, 3H), 6.97 (d, *J* = 8.9 Hz, 2H), 3.86 (s, 3H); <sup>19</sup>F NMR:  $\delta$  71.05(s); IR (neat, cm<sup>-1</sup>): 2942, 2365, 1612, 1587, 1449, 1259, 1154; MS (ESI) *m/z* 304 (M + H<sup>+</sup>); Anal. Calcd for C<sub>17</sub>H<sub>12</sub>F<sub>3</sub>NO: C, 67.32; H, 3.99; N, 4.62. Found: C, 67.36; H, 3.97; N, 4.61%.

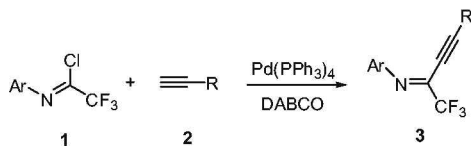
*4-Methoxy-N-(1,1,1-trifluoro-5-phenylpent-3-yn-2-ylidene)aniline (3bb)*: Yellowish solid m.p. 88–90 °C <sup>1</sup>H NMR:  $\delta$  8.12 (d, *J* = 9.3 Hz, 1H), 7.47 (s, 1H), 7.33–7.42 (m, 4H), 7.20–7.31 (m, 3H), 4.41 (s, 2H), 3.85 (s, 3H); <sup>19</sup>F NMR:  $\delta$  71.18(s); IR (KBr, cm<sup>-1</sup>): 3043, 2921, 2201, 1605, 1515, 1461, 1363, 1154; MS (ESI) *m/z* 318 (M + H<sup>+</sup>); Anal. Calcd for C<sub>18</sub>H<sub>14</sub>F<sub>3</sub>NO: C, 68.13; H, 4.45; N, 4.41. Found: C, 68.09; H, 4.46; N, 4.39%.

*4-Methoxy-N-(1,1,1-trifluoro-3-yn-2-ylidene)aniline (3bc)*: Yellowish solid, m.p. 76–78 °C <sup>1</sup>H NMR:  $\delta$  7.42 (d, *J* = 8.9 Hz, 2H), 6.92 (d, *J* = 8.9 Hz, 2H), 3.83 (s, 3H), 2.41 (t, *J* = 7.0 Hz, 2H), 1.54–1.59 (m, 2H), 1.33–1.36 (m, 2H), 1.30–1.32 (m, 2H), 0.96 (t, *J* = 6.8 Hz, 3H); <sup>19</sup>F NMR:  $\delta$  67.37(s); IR (KBr, cm<sup>-1</sup>): 3043, 2958, 2863, 2231,

**Table 2** The Sonogashira reaction of *N*-aryltrifluoroacetimidoyl chlorides and 1-alkyne

Entry	Ar(1a–e)	R(2a–d)	Product 3	Yield/% <sup>a</sup>
1	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> (1a)	Ph(2a)	3aa	93
2	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> (1a)	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> (2b)	3ab	95
3	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> (1a)	<i>n</i> -C <sub>4</sub> H <sub>9</sub> (2c)	3ac	94
4	4-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> (1a)	<i>n</i> -C <sub>5</sub> H <sub>11</sub> (2d)	3ad	96
5	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> (1b)	Ph(2a)	3ba	95
6	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> (1b)	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> (2b)	3bb	94
7	4-CH <sub>3</sub> O-C <sub>6</sub> H <sub>4</sub> (1b)	<i>n</i> -C <sub>5</sub> H <sub>11</sub> (2c)	3bc	87
8	4-Cl-C <sub>6</sub> H <sub>4</sub> (1c)	Ph(2a)	3ca	88
9	4-Cl-C <sub>6</sub> H <sub>4</sub> (1c)	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> (2b)	3cb	73
10	4-Cl-C <sub>6</sub> H <sub>4</sub> (1c)	<i>n</i> -C <sub>4</sub> H <sub>9</sub> (2c)	3cc	83
11	4-Cl-C <sub>6</sub> H <sub>4</sub> (1c)	<i>n</i> -C <sub>5</sub> H <sub>11</sub> (2d)	3cd	82
12	4-Cl-C <sub>6</sub> H <sub>4</sub> (1c)	<i>n</i> -C <sub>3</sub> H <sub>7</sub> (2e)	3ce	83
13	2-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> (1d)	Ph(2a)	3da	88
14	2-CH <sub>3</sub> -C <sub>6</sub> H <sub>4</sub> (1d)	C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> (2b)	3db	81
15	2-NO <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> (1e)	Ph(2a)	3ea	60

<sup>a</sup>Isolated yield.



1593, 1505, 1470, 1373, 1152; MS (ESI)  $m/z$  298 (M + H<sup>+</sup>); Anal. Calcd for C<sub>16</sub>H<sub>18</sub>F<sub>3</sub>NO: C, 64.64; H, 6.10; N, 4.71. Found: C, 64.69; H, 6.13; N, 4.69%.

**4-Chloro-N-(1,1,1-trifluoro-4-phenylbut-3-yn-2-ylidene)aniline (3ca):** Yellowish oil. <sup>1</sup>H NMR: δ 7.42–7.47 (m, 3H), 7.32–7.39 (m, 4H), 7.24–7.29 (m, 2H); <sup>19</sup>F NMR: δ 71.66(s); IR (neat, cm<sup>-1</sup>): 3052, 2926, 2199, 1603, 1507, 1442, 1154; MS (ESI)  $m/z$  308 (M + H<sup>+</sup>); Anal. Calcd for C<sub>16</sub>H<sub>9</sub>ClF<sub>3</sub>N: C, 62.45; H, 2.95; N, 4.55. Found: C, 62.41; H, 2.96; N, 4.57%.

**4-Chloro-N-(1,1,1-trifluoro-5-phenylpent-3-yn-2-ylidene)aniline (3cb):** Yellowish solid, m.p. 106–108°C <sup>1</sup>H NMR: δ 8.14 (d,  $J = 9.0$  Hz, 1H), 8.05 (s, 1H), 7.70 (d,  $J = 9.0$  Hz, 1H), 7.45 (s, 1H), 7.15–7.32 (m, 5H), 4.41 (s, 2H); <sup>19</sup>F NMR: δ 71.83(s); IR (KBr, cm<sup>-1</sup>): 3054, 2918, 2196, 1605, 1516, 1443, 1154; MS (ESI)  $m/z$  322 (M + H<sup>+</sup>); Anal. Calcd for C<sub>17</sub>H<sub>11</sub>ClF<sub>3</sub>N: C, 63.46; H, 3.45; N, 4.35. Found: C, 63.42; H, 3.43; N, 4.36%.

**4-Chloro-N-(1,1,1-trifluoro-3-yn-2-ylidene)aniline (3cc):** Yellowish solid, m.p. 95–97°C <sup>1</sup>H NMR: δ 8.16 (d,  $J = 9.0$  Hz, 1H), 8.06 (s, 1H), 7.75 (d,  $J = 9.0$  Hz, 1H), 7.58 (s, 1H), 3.10 (t,  $J = 8.0$  Hz, 2H), 1.74–1.82 (m, 2H), 1.43–1.53 (m, 2H), 1.01 (t,  $J = 7.3$  Hz, 3H); <sup>19</sup>F NMR: δ 67.63(s); IR (KBr, cm<sup>-1</sup>): 3042, 2961, 2219, 1621, 1599, 1495, 1376, 1154; MS (ESI)  $m/z$  288 (M + H<sup>+</sup>); Anal. Calcd for C<sub>14</sub>H<sub>13</sub>ClF<sub>3</sub>N: C, 58.44; H, 4.55; N, 4.87. Found: C, 58.48; H, 4.53; N, 4.85%.

**4-Chloro-N-(1,1,1-trifluoronon-3-yn-2-ylidene)aniline (3cd):** Yellowish solid, m.p. 96–98°C <sup>1</sup>H NMR: δ 8.15 (d,  $J = 9.0$  Hz, 1H), 8.05 (s, 1H), 7.73 (d,  $J = 9.0$  Hz, 1H), 7.58 (s, 1H), 3.10 (t,  $J = 8.0$  Hz, 2H), 1.76–1.83 (m, 2H), 1.47–1.60 (m, 2H), 1.37–1.46 (m, 2H), 0.93 (t,  $J = 7.1$  Hz, 3H); <sup>19</sup>F NMR: δ 67.65(s); IR (KBr, cm<sup>-1</sup>): 3042, 2931, 2217, 1599, 1509, 1495, 1375, 1154; MS (ESI)  $m/z$  302 (M + H<sup>+</sup>); Anal. Calcd for C<sub>15</sub>H<sub>15</sub>ClF<sub>3</sub>N: C, 59.71; H, 5.01; N, 4.64. Found: C, 59.68; H, 5.02; N, 4.62%.

**4-Chloro-N-(1,1,1-trifluorohept-3-yn-2-ylidene)aniline (3ce):** Yellowish solid, m.p. 131–133°C <sup>1</sup>H NMR: δ 8.15 (d,  $J = 9.0$  Hz, 1H), 8.05 (s, 1H), 7.73 (d,  $J = 9.0$  Hz, 1H), 7.58 (s, 1H), 3.07 (t,  $J = 7.6$  Hz, 2H), 1.79–1.87 (m, 2H), 1.07 (t,  $J = 7.4$  Hz, 3H); <sup>19</sup>F NMR: δ 67.67(s); IR (KBr, cm<sup>-1</sup>): 3042, 2961, 2299, 1691, 1599, 1495, 1376, 1154; MS (ESI)  $m/z$  274 (M + H<sup>+</sup>); Anal. Calcd for C<sub>13</sub>H<sub>11</sub>ClF<sub>3</sub>N: C, 57.05; H, 4.05; N, 5.12. Found: C, 57.09; H, 4.07; N, 5.09%.

**2-Methyl-N-(1,1,1-trifluoro-4-phenylbut-3-yn-2-ylidene)aniline (3da):** Yellowish oil. <sup>1</sup>H NMR: δ 7.42–7.55 (m, 3H), 7.24–7.36 (m, 4H), 7.16–7.19 (m, 2H), 2.83 (s, 3H); <sup>19</sup>F NMR: δ 71.17(s); IR (neat, cm<sup>-1</sup>): 3062, 2934, 2200, 1607, 1479, 1347, 1150; MS (ESI)  $m/z$  288 (M + H<sup>+</sup>); Anal. Calcd for C<sub>17</sub>H<sub>12</sub>F<sub>3</sub>N: C, 71.07; H, 4.21; N, 4.88. Found: C, 71.04; H, 4.18; N, 4.87%.

**4-Methyl-N-(1,1,1-trifluoro-5-phenylpent-3-yn-2-ylidene)aniline (3db):** Yellowish oil <sup>1</sup>H NMR: δ 7.88 (d,  $J = 8.2$  Hz, 1H), 7.57 (d,  $J = 8.2$  Hz, 1H), 7.45 (t,  $J = 6.4$  Hz, 2H), 7.12–7.30 (m, 5H), 4.43 (s, 2H), 2.82 (s, 3H); <sup>19</sup>F NMR: δ 71.26(s); IR (neat, cm<sup>-1</sup>): 3065, 2962, 2214, 1619, 1467, 1336, 1150; MS (ESI)  $m/z$  302 (M + H<sup>+</sup>); Anal. Calcd for C<sub>18</sub>H<sub>14</sub>F<sub>3</sub>N: C, 71.75; H, 4.68; N, 4.65. Found: C, 71.72; H, 4.70; N, 4.66%.

**2-Nitro-N-(1,1,1-trifluoro-4-phenylbut-3-yn-2-ylidene)aniline (3ea):** Yellowish oil. <sup>1</sup>H NMR: δ 7.53–7.54 (m, 4H), 7.26–7.36 (m, 5H); <sup>19</sup>F NMR: 71.70(s); IR (neat, cm<sup>-1</sup>): 3065, 2922, 2236, 1591, 1479, 1152; MS (ESI)  $m/z$  319 (M + H<sup>+</sup>). Anal. Calcd for C<sub>16</sub>H<sub>9</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub>: C, 60.38; H, 2.85; N, 8.80. Found: C, 60.35; H, 2.83; N, 8.83.

We thank the Office of Education of Zhejiang Province (20070535) and Wenzhou Municipal Science and Technology Bureau (G20060075) for financial supports.

Received 26 June 2008; accepted 8 September 2008

Paper 08/0024 doi: 10.3184/030823408X374224

Published online: 30 October 2008

## References

- 1 B.E. Smart, *J. Fluorine Chem.*, 2001, **109**, 3.
- 2 D. Gimenez, C. Andreu and M. Olmo, et al. *Bioorg. Med. Chem.*, 2006, **14**, 6971.
- 3 G. Magueur, B. Crousse, S. Charneau, P. Grellierm, J.-P. Begue and D. Bonnet-Delpon, *J. Med. Chem.*, 2004, **47**, 2694.
- 4 A. Gryshuk, Y. Chen and L.N. Goswami, et al. *J. Med. Chem.*, 2007, **50**, 1754.
- 5 V.N. Korotchenko, A.V. Shastin, V.G. Nenajdenko and E.S. Balenkova, *Tetrahedron*, 2001, **57**, 7519.
- 6 G.K. Prakash, J. Hu and G.A. Olah, *Org. Lett.*, 2003, **5**, 3253.
- 7 H. Amii, Y. Kishikawa and K. Uneyama, *Org. Lett.*, 2001, **3**, 1109.
- 8 H.B. Yu and W.Y. Huang, *J. Fluorine Chem.*, 1998, **87**, 69.
- 9 K. Komeyama, D. Sasayama, T. Kawabata, K. Takehira and K. Takaki, *J. Org. Chem.*, 2005, **70**, 10679.
- 10 G. Magueur, B. Crousse and D. Bonnet-Delpon, *Tetrahedron Lett.*, 2005, **46**, 2219.
- 11 A.V. Kel'in, A.W. Sromek and V.J. Gevorgyan, *J. Am. Chem. Soc.*, 2001, **123**, 2074.
- 12 B.G. Van den Hoven and H. Alper, *J. Am. Chem. Soc.*, 2001, **123**, 10214.
- 13 R. Faust and B. Gobelt, *Tetrahedron Lett.*, 1997, **38**, 8017.
- 14 S.Y. Liu, H.Y. Sheng and Y.Z. Huang, *Synthesis*, 1991, 235.
- 15 Z. Gu, Z.Z. Li, Z.C. Liu, Y. Wang, C.B. Liu and J.N. Xiang, *Catal. Commun.*, 2008, **9**, 2154.
- 16 B. Liang, M.J. Dai, J.H. Chen and Z. Yang, *J. Org. Chem.*, 2005, **70**, 391.
- 17 C.Y. Yi and R.M. Hua, *J. Org. Chem.*, 2006, **71**, 2535.
- 18 K. Uneyama and H. Watanabe, *Tetrahedron Lett.*, 1991, **32**, 1459.
- 19 Y.M. Wu, M. Zhang and Y.Q. Li, *J. Fluorine Chem.*, 2006, **127**, 218.